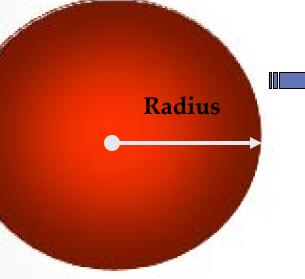
Chemistry Vocabulary

Atomic Radius



Atomic radius is the distance from the center of an atom's nucleus to its outermost electron.

Isotopes Atoms of the same element (same atomic

number) with **different** mass numbers

Isotopes of chlorine ³⁵Cl ³⁷Cl ¹⁷ ¹⁷ chlorine - 35 ³⁷Cl ¹⁷ ³⁷Cl ¹⁷ ³⁷Cl ¹⁷ ³⁷Cl ¹⁷ ³⁷Cl ³⁷Cl

Alpha Decay

Parent nuclide

238

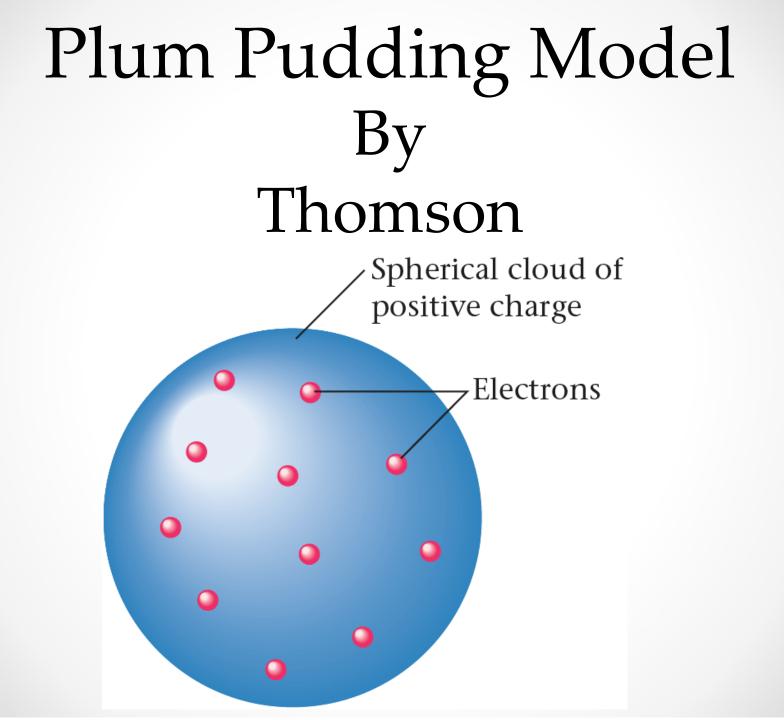
Daughter nuclides

 $^{234}_{90}$ Th + $^{4}_{2}$ He

1A 1	Valence Electrons					8A ² He
H٠						1s ²
1s ¹	2A	3A	4A	5A	6A	7A
3	4	5	6	7	8	9
Li	•Be•	۰B۰	٠Ċ٠	٠Ņ٠	:Ö·	۰F۰
[He] 2s ¹	[He] 2s ²	[He] 2s ² 2p ¹	[He] 2s ² 2p ²	[He] 2s ² 2p ³	[He] 2s ² 2p ⁴	[He] 2s ² 2p ⁵

Number of valence electrons is equal to the Group number.

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Chemical Change A change in which one or more

substances are converted into different substances.



Physical Change

• A change that occurs that does not change the identity of the substance.

Melting ice(change in state or phase)
Tearing paper



Linear

2 atoms attached to center atom

0 unshared pairs (lone pairs)

Bond angle = 180°

Ex. : BeF_2

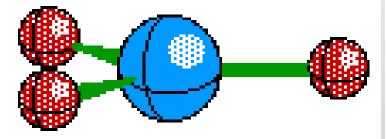


Trigonal Planar

- 3 atoms attached to center atom
- 0 lone pairs

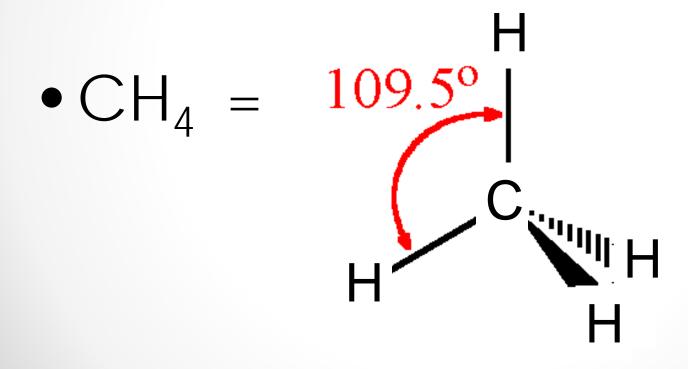
• Bond angle = 120°

• Ex. : AlF_3



Tetrahedral

• 4 atoms bonded together.

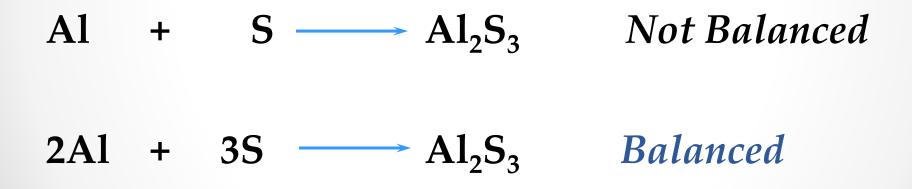


Law of Conservation of Mass

In any ordinary chemical reaction, matter is not created nor destroyed.

Balanced Chemical Equation

Same numbers of each type of atom on each side of the equation



EXOTHERMIC

A change (e.g. a chemical reaction) that releases heat.

A release of heat corresponds to a decrease in enthalpy

Burning fossil fuels is an exothermic reaction



Endothermic

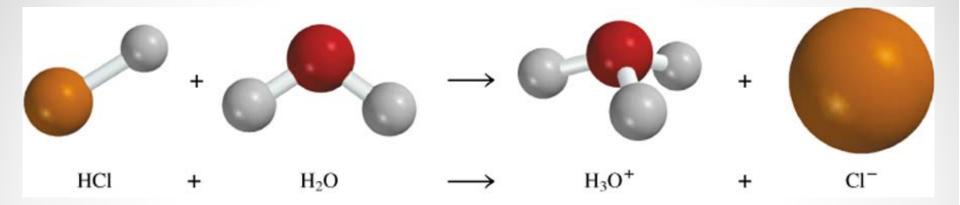
• A change (e.g. a chemical reaction) that requires (or absorbs) heat.

• An input of heat corresponds to an increase in enthalpy

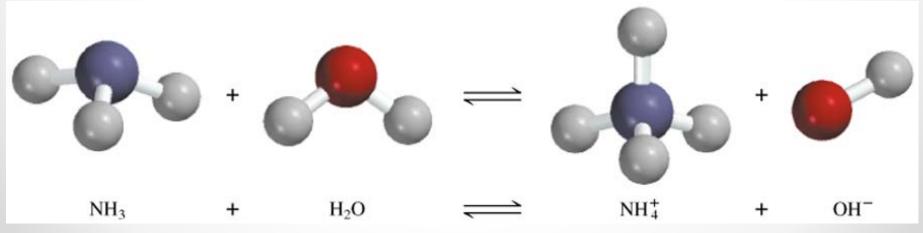
Photosynthesis is an endothermic reaction (requires energy input from sun)



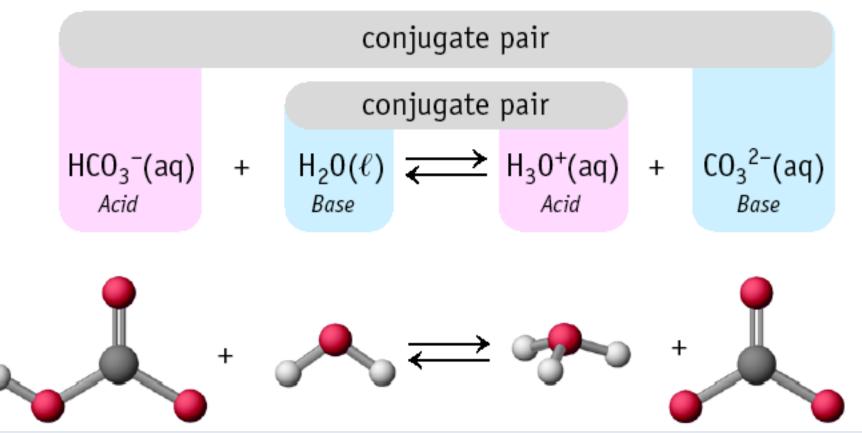
Arrhenius acid is a substance that produces $H^+(H_3O^+)$ in water



Arrhenius base is a substance that produces OH⁻ in water



Conjugate Pairs



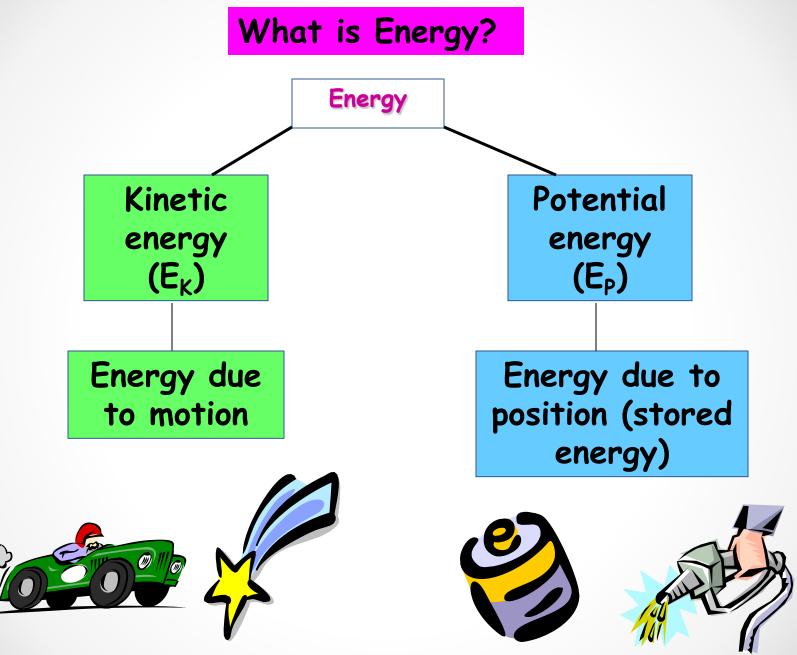
Cohesion







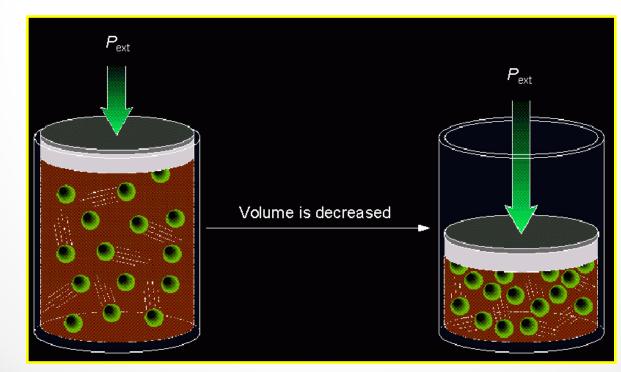
Helps insects walk across water •



Boyle's Law

Gas <u>pressure is inversely proportional to the</u> <u>volume</u>, when temperature is held constant.

• Equation: $P_1V_1 = P_2V_2$ (T = constant)



Solubility Curve

